Application note (AN-2019-01EN)

CO₂ breakthrough curve measurement using BELCATII

Introduction

The Adsorption breakthrough curve measurement is widely used as an assay method to examine design parameters and adsorption rates for adsorption process. In this article, aiming at single component gas recovery of CO_2 , which is one of the greenhouse gases. we conducted CO_2 breakthrough curve measurement, and at the same time, helium purge and TPD measurement to observe the regeneration treatment process.



Experimental

As an adsorbent, zeolite molecular sieve 5A was filled in a triple structure sample tube of BELCATII, pretreated at 400 °C in helium flow, and a breakthrough curve measurement was carried out in 1% - CO₂/He (50 SCCM) gas flow. After this, the sample tube was purged with helium as a reactivation treatment, and TPD-measurement (temperature programmed desorption) was performed. The same measurement was performed for an empty sample cell. The adsorption/desorption amount in each process and the mass balance in the entire measurement were evaluated from the difference of each profile. The built-in thermal conductivity detector (TCD) was used as a detector.

Adsorbent: Pretreatment: Breakthrough curve measurement: He purge: TPD measurement:

Zeolite molecular sieve 5A (Amount: 0.1 g, Particle size: 250 to 500 μ m) In 100% He flow (50 SCCM) At 400 °C, for 60 minutes. In 1% CO₂/He (50 SCCM) flow, at 25 °C, for 25 minutes. In 100% He (50 SCCM) flow, at 25 °C, for 50 minutes. In 100% He (50 SCCM) flow, at a 10°C min⁻¹ ramp from 25 °C to 200 °C.

Result and discussion

The continuous breakthrough curve – TPD measurement result are shown in Figure 1. The break point and the end point were observed about 5 minutes, and 10 minutes respectively from starting point of the breakthrough curve measurement. On the other hand, it took 50 minutes for the reactivation process, and about 10% of CO₂ was desorbed during the TPD measurement. It is considered that there might be some strong adsorption sites on MS-5A, on which CO₂ molecules can be adsorbed strongly.

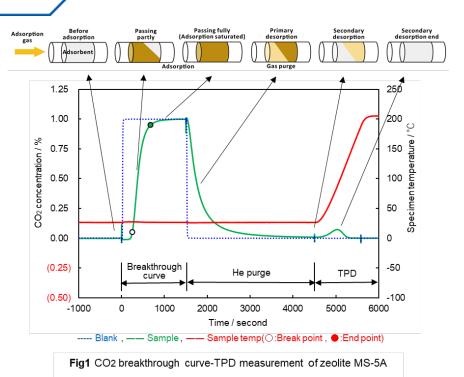


Figure 2 shows the difference between a blank sample measurement and a sample measurement result. From the shaded area of each peak, the adsorption/desorption amount can be calculated (Table 1).

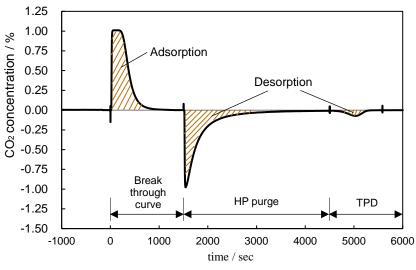


Fig.2 Difference between blank and sample measurement

| Table 1: Amount of adsor | ption and desor | ption in each process |
|--------------------------|-----------------|-----------------------|
| | puon ana acsor | ption in cuch process |

| | Breakthrough curve(Ads.) | He purge(Des.) | TPD(Des.) | | |
|--|---|-----------------------|-------------------------|--|--|
| | Ads.amount: 1.41 mmol/g | Des.amount: 1.26 mmol | Des.amount: 0.12 mmol/g | | |
| | Break point (Saturation degree 5%): 250 seconds | | Peak top tamp.: 90°C | | |
| | End point (Saturation degree 95%): 680 seconds | | | | |
| | Mass balance : (1.26+0.12) / 1.41×100 = 97.9% | | | | |

Thus, by the continuous breakthrough – He purge – TPD measurement, the adsorption amount and desorption amount (reactivation process) of adsorbents can be evaluated quantitively.